



Maturitní témata

Školní rok:			2022/2023	
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Předmětová komise:			Biologie Chemie	
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Schváleno předmětovou komisí dne:			29.8.2022	Podpis:
Schváleno ředitelem školy dne:				Podpis a razítko:
Počet výtisků:	3	Výtisk č.:		

1. Periodic system of elements

- periodic law
- trends across the periodic system of elements
- uniqueness of noble gases within the periodic system of elements
- s, p, d, f blocks characterisation
- principle of a chemical bond, types of bonds

2. Atomic structure

- models of atomic structure
- quantum numbers, electron configuration
- electronic structural formulae
- VSEPR theory interaction of lone electron pairs

3. Sources of energy

- enthalpy change
- thermochemical equations physical states of matter
- combustion vs. nuclear energy
- fossil fuels combustion of alkanes
- cellular respiration vs. photosynthesis

4. Catalysts

- chemical kinetics main assumptions of the collision theory
- a catalyst -an important factor affecting the reaction rate
- catalysts used in inorganic and organic syntheses and processes
- biological catalysts enzyme kinetics

5. Equilibrium

- dynamic equilibrium
- Guldberg- Waag expression
- factors affecting equilibrium
- pH ionic product of water and its derivation, calculations, uses

6. Analytical chemistry

- qualitative analysis detection reactions of inorganic anions and cations, detection reactions in organic chemistry
- quantitative analysis neutralisation and redox titrations
- acid base theory
- strong and weak electrolytes

7. Atmosphere

- components of the air, separating the mixtures
- chemical properties of oxygen, nitrogen and noble gases
- environmental problems pollution of the air and its consequences
- importance of the process of photosynthesis

8. Water

- unique properties of water
- hydrogen bridges
- water hardness
- processes leading to potable water, eutrophication of water
- hydrolysis

9. Oxygen

- oxygen occurrence, manufacture, properties, uses
- oxides and peroxides, oxidation numbers
- ethers important characteristics, preparation, properties, uses
- comparison of H₂O, R-OH and R-O-R

10. Hydrogen

- hydrogen occurrence, manufacture, properties, uses, isotopes
- hydrogenation important chemical process, hardening of fats
- unsaturated hydrocarbons

11. Important non-metals

- characteristic properties of non-metals
- sulphur occurrence, properties
- production of sulphuric (VI) acid
- glass manufacture importance of silicon

12. Properties of metals

- characteristic properties of metals
- chemical behaviour of metals reactivity series
- p- block metals
- s- block metals compounds of I.A and II.A metals
- ionic bond

13. Transition metals

- characterisation of the d- block elements electron configurations and their exceptions
- complex compounds and dative bond
- iron properties, manufacture, uses
- the role of iron in the transport of oxygen haem group characterisation of heterocyclic compounds
- transition metals as enzyme inhibitors types of inhibition

14. Sources of organic compounds

- fossil fuels their importance, uses and risks
- fractional distillation of crude oil
- classification of hydrocarbons
- saturated hydrocarbons

15. Isomerism

- allotropic modifications
- shapes of molecules
- structural isomerism and stereoisomerism
- properties of alkanes, alkenes, amino acids and saccharides with respect to their typical isomerism

16. Compounds of halogens

- elements of the VII. A group trends, physical and chemical properties, compounds, uses
- organic halogen compounds
- polarity of a covalent bond
- environmental problems freons

17. Compounds of nitrogen

- chemistry of nitrogen physical and chemical properties, compounds, uses
- properties of nitrogen compared to properties of phosphorus
- organic nitrogen compounds amines, nitro compounds, nitration

18. Hydroxycompounds

- physical and chemical properties of hydroxycompounds
- differences between alcohols and phenols
- fermentation
- separating mixtures
- terpenes

19. Carbonyl compounds

- physical and chemical properties of carbonyl compounds
- differences between aldehydes and ketones
- typical chemical reactions
- monosaccharides polyhydroxyaldehydes, polyhydroxyketones
- redox properties reduction, oxidation

20. Proteins

- amino acids the building blocks of proteins
- characterisation of substitutional derivatives of carboxylic acids
- nitrogen an important bioelement
- different functions of proteins
- denaturing

21. Lipids

- different functions of lipids
- oils and fats esters of glycerol and fatty acids
- characterisation of functional derivatives of carboxylic acids
- biological importance of steroids

22. Genetic information

- nucleic acids DNA, RNA
- phosphorus an important bioelement
- protein synthesis
- genetic engineering

23. Metabolism

- thermochemical laws
- anabolic vs. catabolic processes
- controlling the metabolic processes enzymes, hormones
- vitamins

24. Macromolecular compounds

- naturally occurring macromolecular compounds
- polysaccharides, proteins, nucleic acids their functions and importance
- synthetic macromolecular compounds addition polymers and condensation polymers

25. Structure and properties

- aromatic hydrocarbons properties of the aromatic ring delocalisation
- reaction mechanisms
- functional groups
- biological importance of intermolecular forces structural levels of proteins, DNA double helix
- allotropic modifications of sulphur, phosphorus and carbon